

Implementierung und Verbreitung von Wasserstofftechnologie: Technologische Fortschritte und Herausforderungen

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HyCentA Research GmbH, Graz 2024

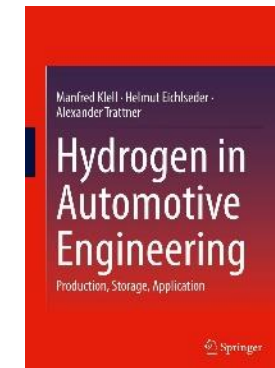
Austria's research center for hydrogen technologies



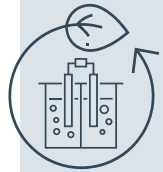
Non-university research organization at
Graz University of Technology (TUG)



- **100+ researchers** from mechanical engineering, physics, chemistry, process engineering, electrical engineering
- More than **600+ projects** successfully completed
- More than **20 years of expertise** State-of-the-art testing & refueling infrastructure
- Teaching at TU Graz
- International network



Research Areas along the Entire Value Chain



Area 1 – ECT Electrolysis and Power-to-X



Material research, new **electrolysis technologies**, alternative processes (from materials to industrial applications)



Area 2 – INT Green Energy and Industry



Research on **generation, purification, compression, storage, distribution and delivery** technologies for the industry and the energy Sector



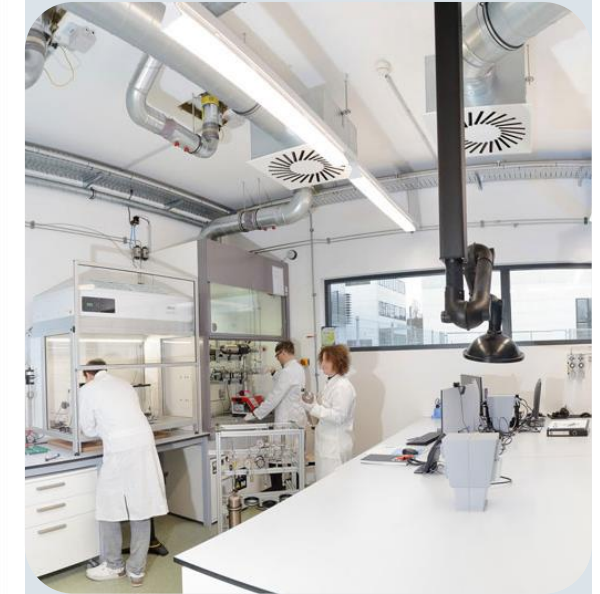
Area 3 – MOT Green Mobility



Fuel cell research on materials, cell, stack and system; optimisation of entire **powertrain** system including **hydrogen storage**



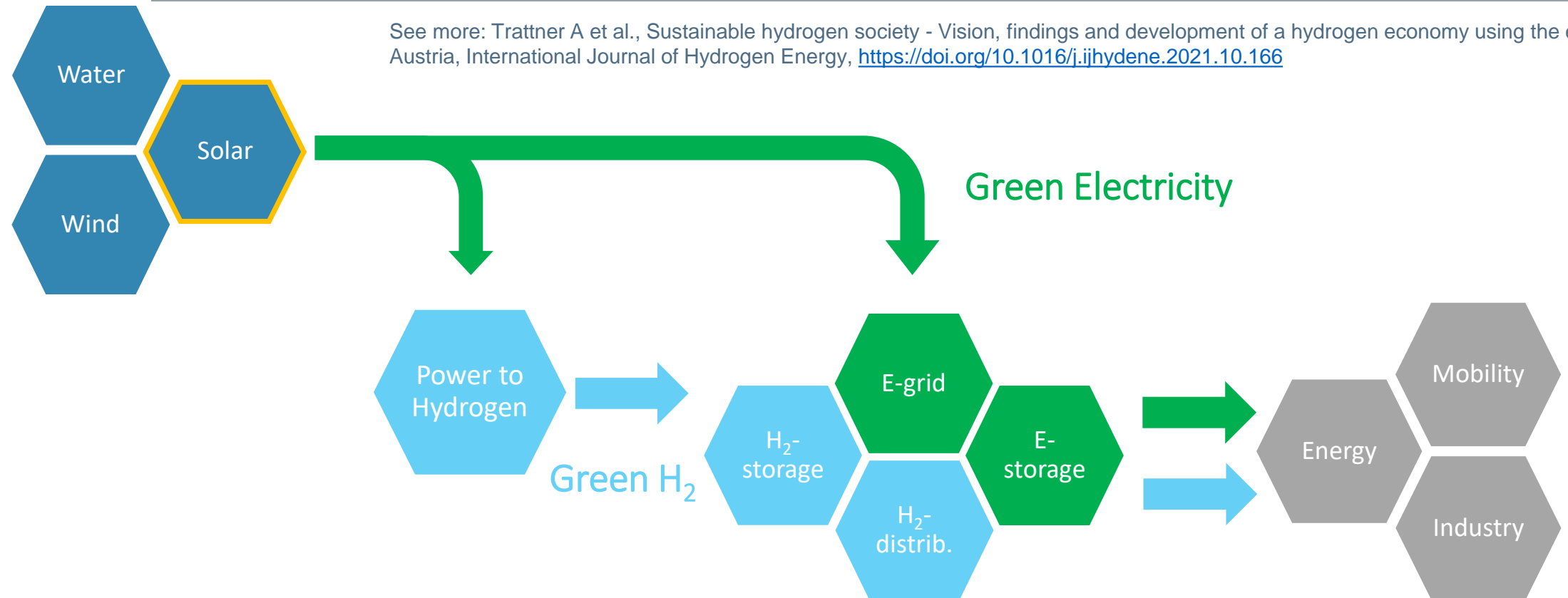
Area 4 – MET Circularity and Sys. Optimization



Measurement and **testing** technologies, **controls**, diagnostics, **modelling** and simulation “digital twin”

Hydrogen – a Key to the Energy Transition

See more: Trattner A et al., Sustainable hydrogen society - Vision, findings and development of a hydrogen economy using the example of Austria, International Journal of Hydrogen Energy, <https://doi.org/10.1016/j.ijhydene.2021.10.166>



Integration of renewables

- Integrate production surpluses
- Direct water splitting

Energy conversion

- Electrolysis - compensate temporal volatility
- H₂ as secondary energy carrier – energy storage

Storage and distribution

- Centralized and decentralized storage
- Long-term storage
- Efficient transport over long distances

Zero Emission Usage

- Energy Services – CHP
- Mobility with Fuel cells
- Industry and high-temperature processes

Grey Hydrogen in AUT – Steam reforming

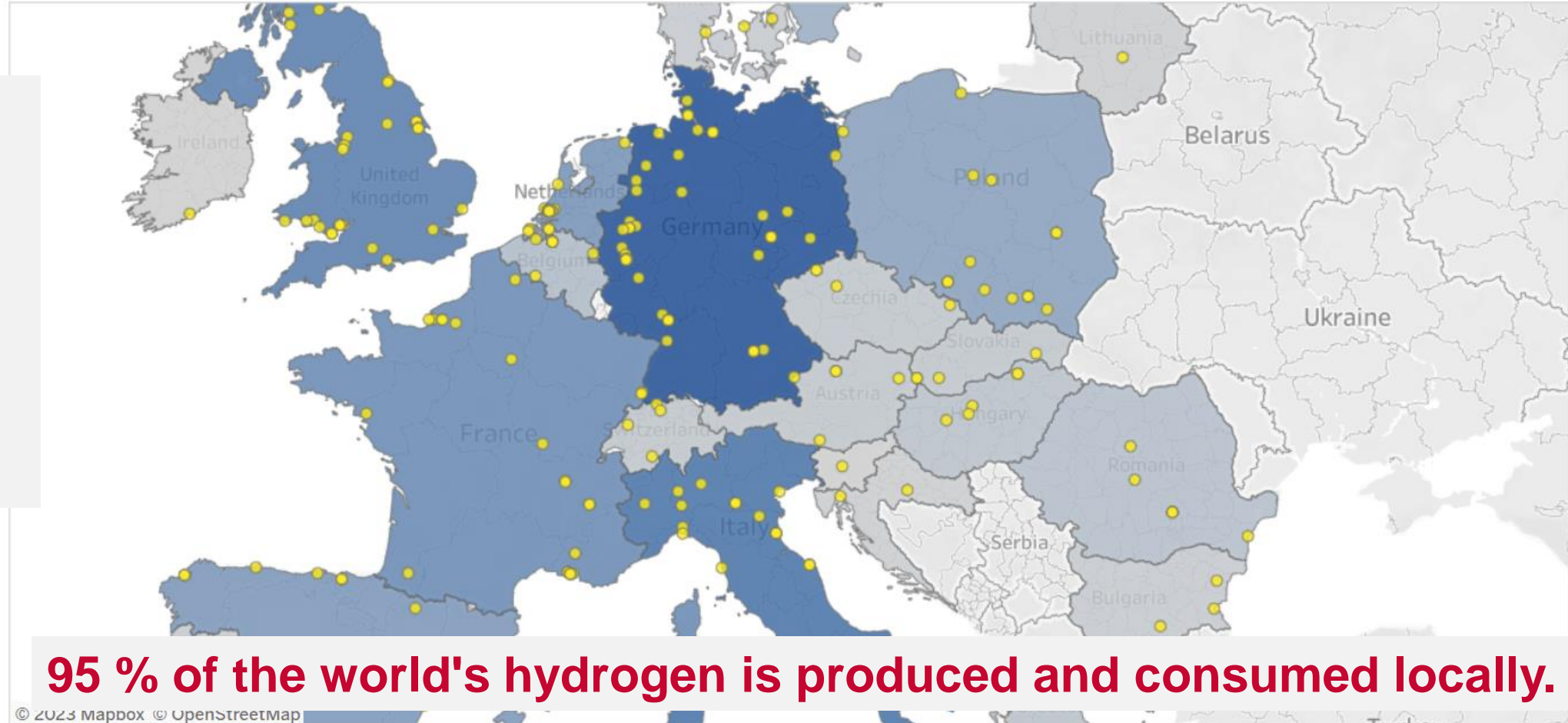
Process / Source

Reforming



2020:
Austria
433 t/day

3 large-scale
steam
reformers



Source: <https://www.fchobservatory.eu>

Number of hydrogen production plants
1 44

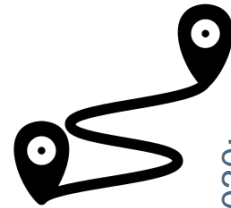
- Currently, the annual hydrogen demand of industry in Austria (primarily in the chemical and petrochemical industry) is **around 140,000 tonnes** (source: H2 Strategy AUT), which is produced from fossil sources (natural gas).
- **140,000 tonnes = 5.6 TWh** (calorific value) → 1.4 % share in the primary energy system (approx. 400 TWh)
- Would correspond to an **electrolysis capacity of approx. 830 MW** (8000 h/a - 70% efficiency)
- **Goal = 1 GW in Austria until 2030**

 **Bundesministerium**
Klimaschutz, Umwelt,
Energie, Mobilität,
Innovation und Technologie

 **Bundesministerium**
Digitalisierung und
Wirtschaftsstandort



Wie kann sich **Europa** und **Österreich** mit **grünem Wasserstoff** versorgen?



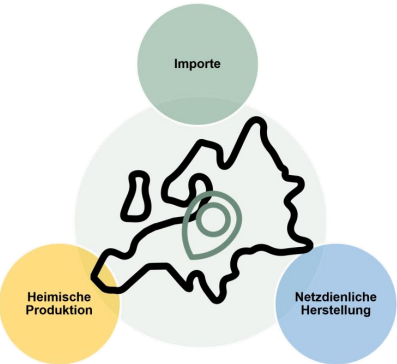
AUT 2030:
20+ TWh

AUT 2030:
5 TWh

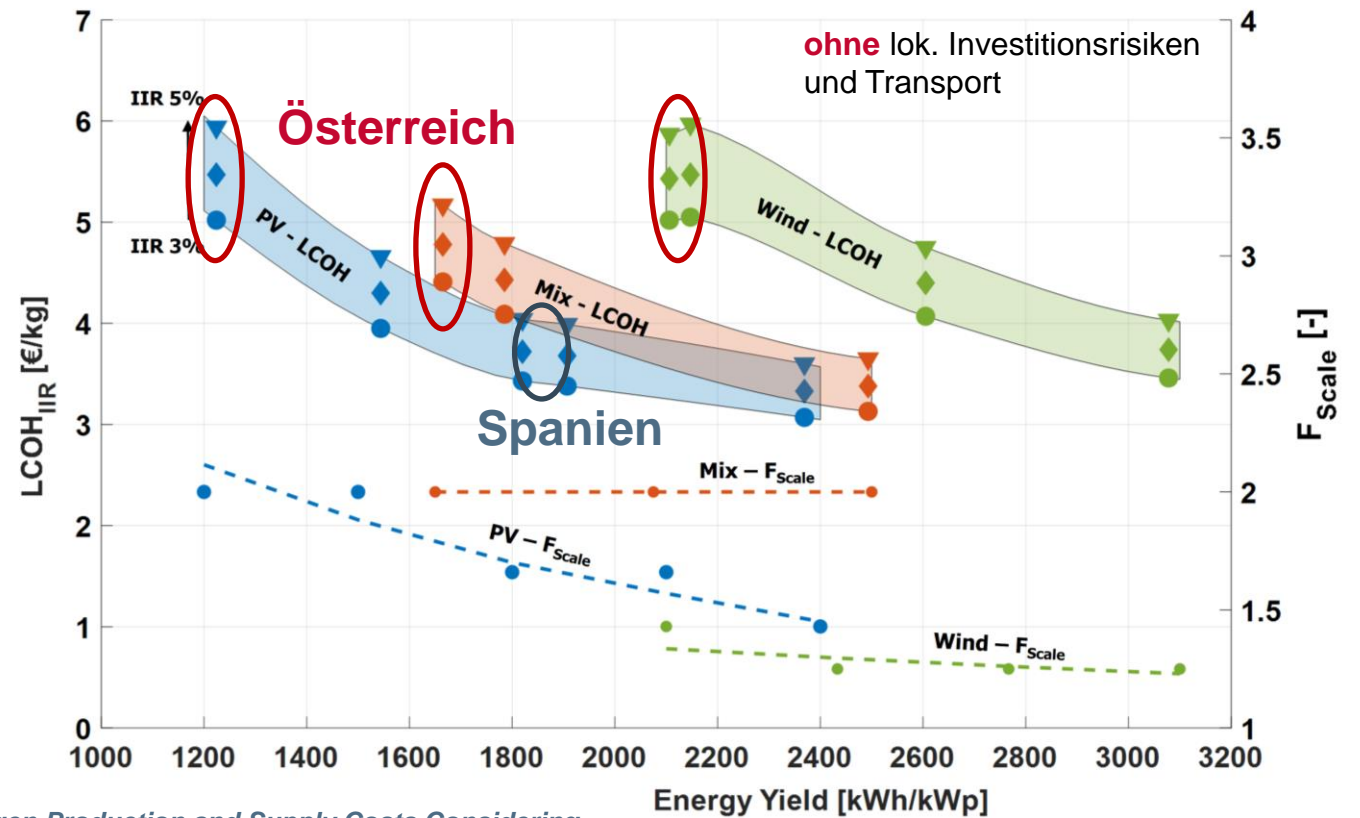


- **Importe:** (Off-Grid) H₂ Produktion im Ausland und Transport nach Europa (Pipeline/Schiff)
 - **Netzdienliche Herstellung:** Nutzen von Überschussenergie zu Stromgestehungskosten
 - **Heimische Produktion:** Nutzen von grüner Energie zur H₂ Produktion lt. RED II
- **Wievil kostet erneuerbarer Wasserstoff?**

H₂-Importe: Berücksichtigen von **lokaler Energieausbeute**, Investitionsrisiken und Transportkosten



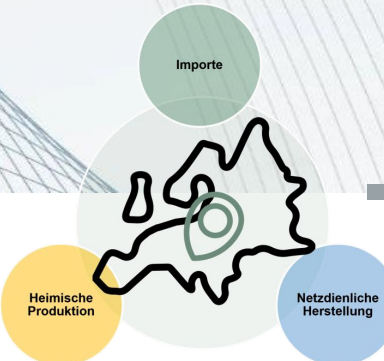
- **Höhere Energieausbeute** – niedrigere LCOH
- **Gemischte Energiequellen** – niedrigere LCOH
- **Leistungspaarung Erneuerbar-Elektrolyse standortabhängig**



Österreich nicht konkurrenzfähig?

Source: Radner, F; Et al. Off-Grid Hydrogen Production: Analysing Hydrogen Production and Supply Costs Considering Country-Specifics and Transport to Europe. IJHE, 2024, <https://doi.org/10.1016/j.ijhydene.2024.07.142>

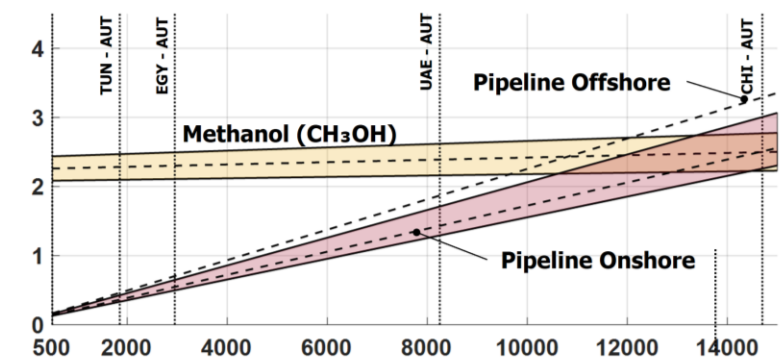
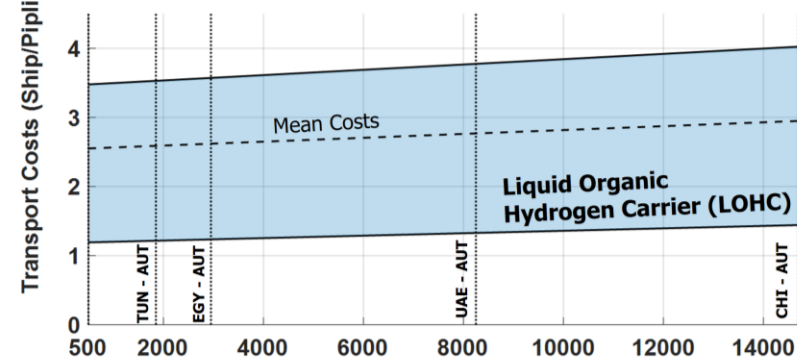
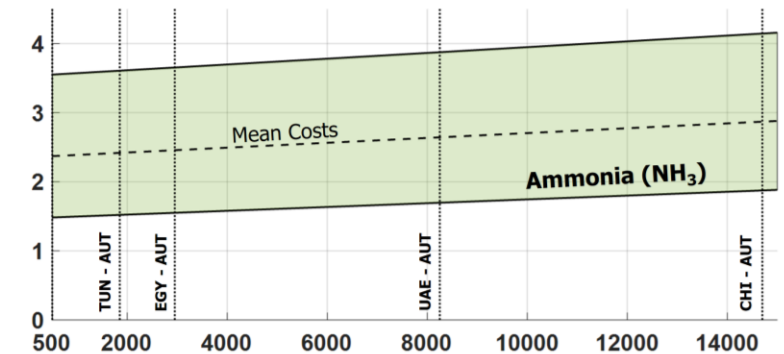
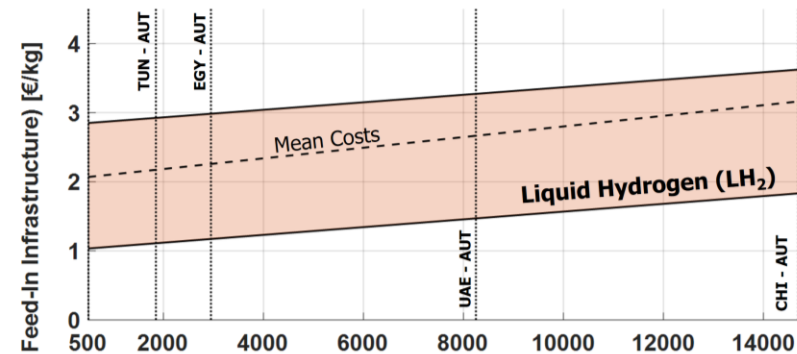
Importierter Wasserstoff



H₂-Importe: Berücksichtigen von lokaler Energieausbeute, Investitionsrisiken und **Transportkosten**

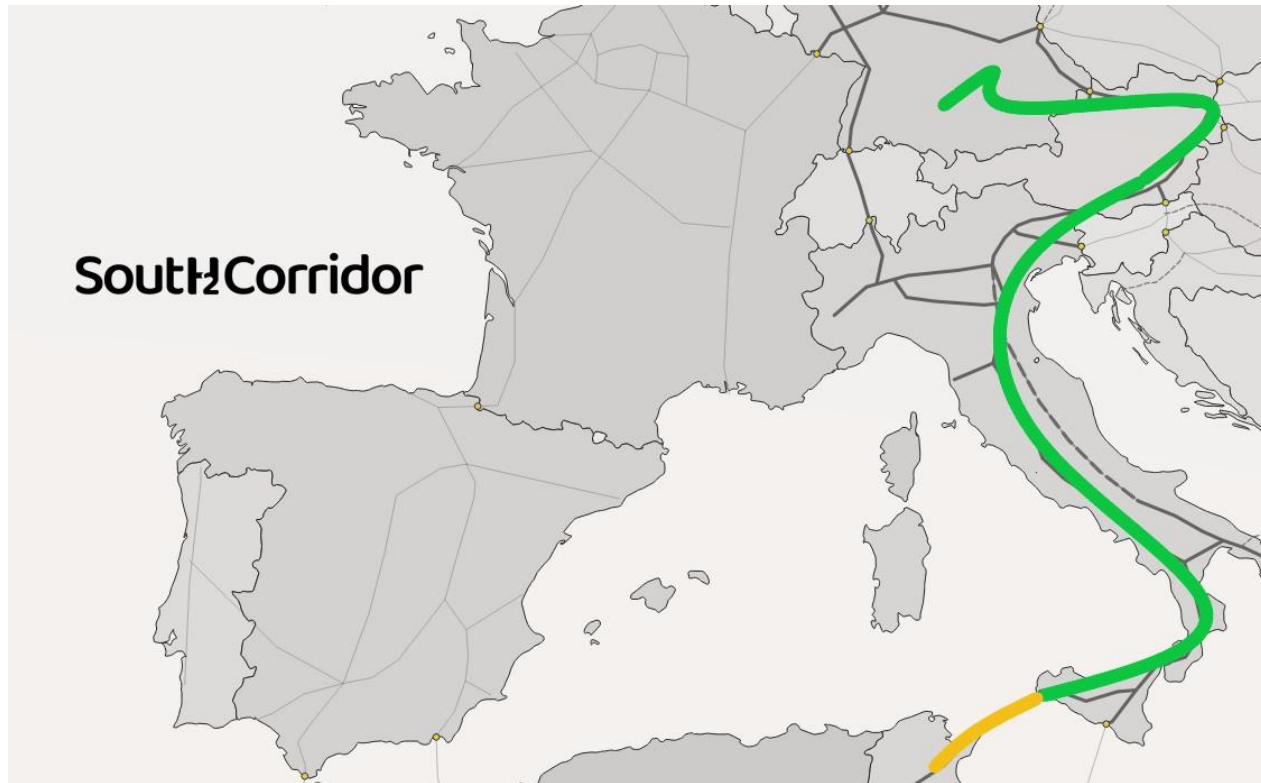
- CHI-AUT: 2,9 - 4,4 €/kg
- UAE-AUT: 1,5 - 4,0 €/kg
- EGY-AUT: 1,3 - 3,8 €/kg
- TUN-AUT: 0,3 - 0,4 €/kg

- **Transportkosten** sind **Entscheidend**
- Form und **Kosten** sind **unklar**
- **Pipeline** am **günstigsten**



Source: Radner, F; Et al. Off-Grid Hydrogen Production: Analysing Hydrogen Production and Supply Costs Considering Country-Specifics and Transport to Europe. IJHE, 2024, <https://doi.org/10.1016/j.ijhydene.2024.07.142>

Fernleitungsnetz (international)

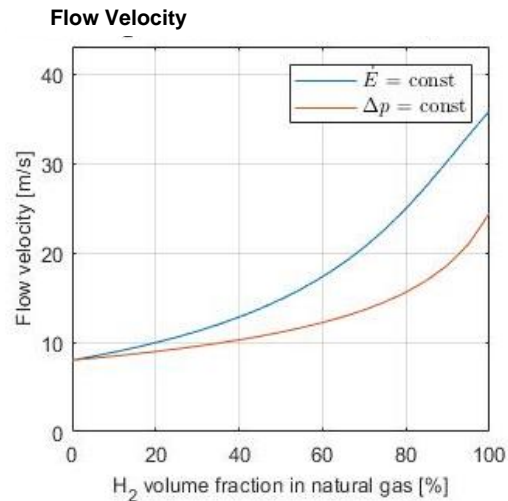
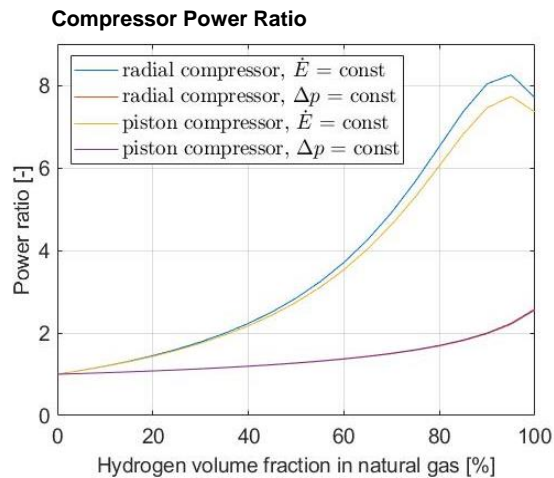
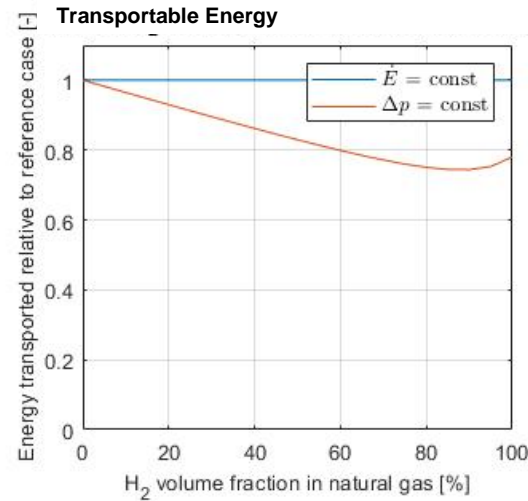


- **South H₂ Corridor is a PCI Status (Project of Common Interest), Funding allowed**
- **Insgesamt 3.300 km Pipeline von Tunesien bis Bayern**
- **70% Umwidmung und 30% Neubau steigt um Faktor 3-4**
- **Potential 4 Mtpa ab 2030**
- **Beabsichtigte Produktionsmenge in Nord Afrika 2030 2.5 Mtpa**

Source: [SouthH2 - Home \(south2corridor.net\)](http://SouthH2 - Home (south2corridor.net))

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H₂ Charakteristik in Pipelines



- **~80% der Energie von Erdgas kann mit H₂ übertragen werden (bei gleichem Δp)**
- **Strömungsgeschwindigkeit steigt um Faktor 3-4**
- **Höhere Strömungsgeschwindigkeit kompensiert die geringere Vol. Energiedichte von H₂**
- **Verdichterleistung steigt um Faktor >2 (bei gleichem Δp)**
- **H₂ Beimischung bis ~25% gangbar**
- **Verringertes Line Pack führt zu **größerer Bedeutung von dezentralen Speicherungen****

Source: Klopčič, N.; Stöhr, T.; Grimmer, I.; Sartory, M.; Trattner, A. Refurbishment of Natural Gas Pipelines towards 100% Hydrogen -A Thermodynamic-Based Analysis. Energies 2022, 15, 9370

H₂ Gas quality: HyGrid Pilot Study

1995, MOP70

1972, MOP70

1965, MOP6

1980 , 2018 MOP70



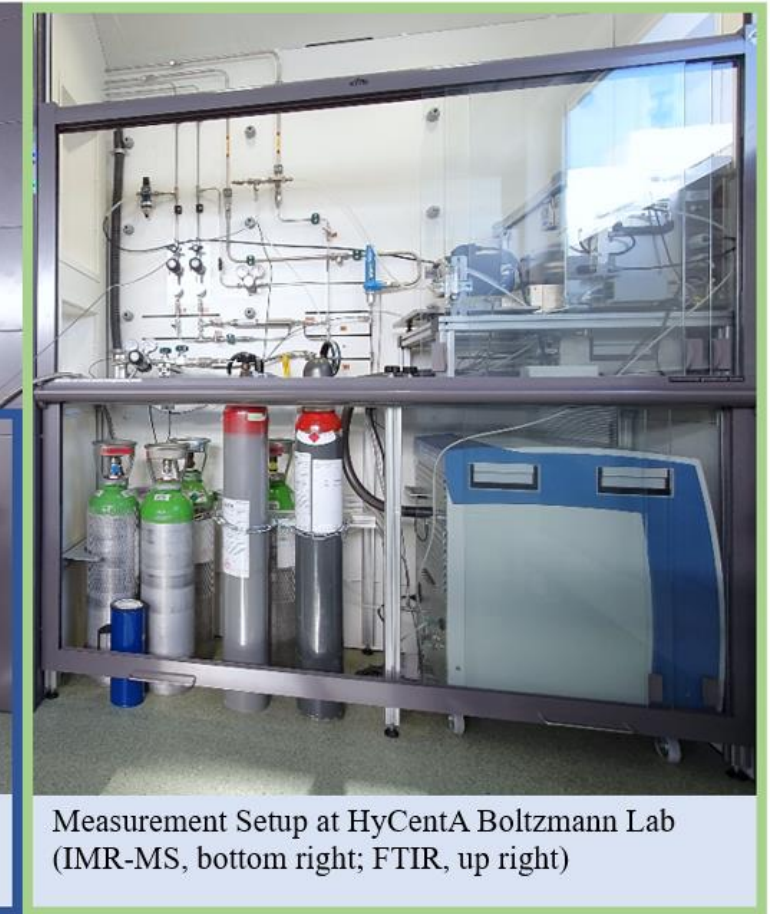
Experimental determination of H2 quality in the Boltzmann Lab

What H2 quality can be achieved during transport in used high-pressure pipes?

- Strong influence of age, gas history, preparation
- Odorants are visible (ppb level!)
- Natural gas residues
- Higher HCs from solid deposits
- Grade A can easily be achieved



Investigated Pipeline Element No. 5



Measurement Setup at HyCentA Boltzmann Lab (IMR-MS, bottom right; FTIR, up right)

AEL

Alkaline Electrolysis

alkaline liquid

PEM-EL

Proton Exchange Membrane Electrolysis

acidic solid

AEM-EL

Anion Exchange Membrane Electrolysis

alkaline solid & liquid

PCC-EL

Proton-Conducting Ceramics Electrolysis

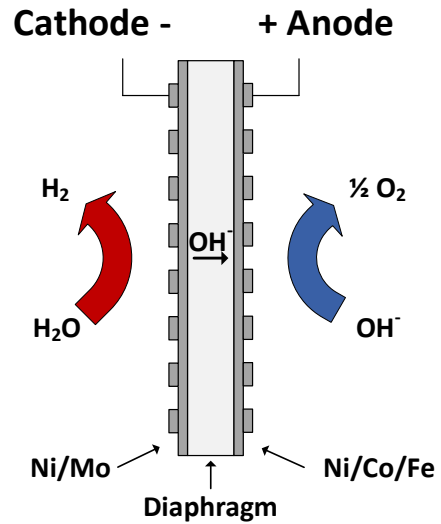
H⁺-conducting ceramic

SO-EL

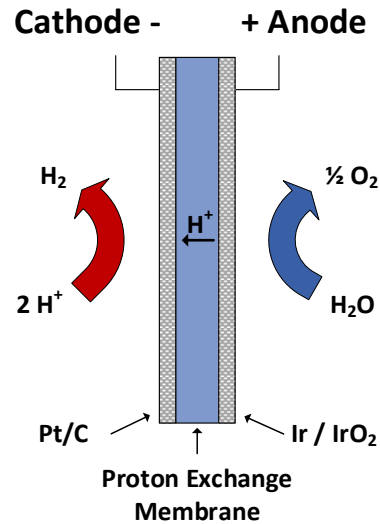
Solid Oxide Electrolysis

O²⁻-conducting ceramic

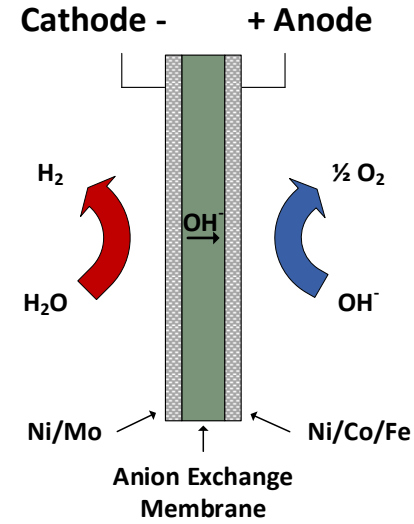
Electrolyte



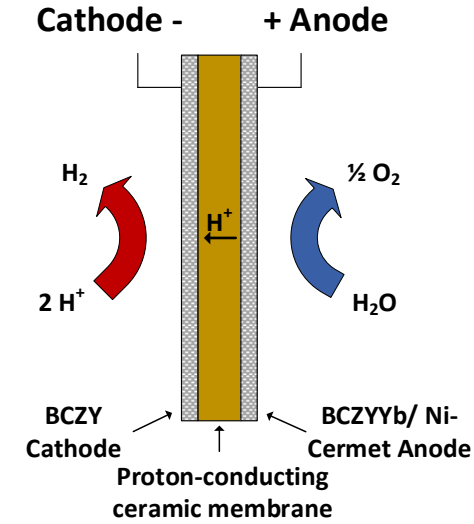
70 - 95 °C



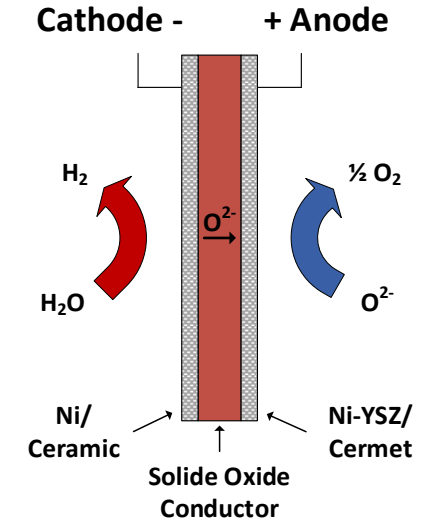
60 - 80 °C



40 - 80 °C



400 - 700 °C



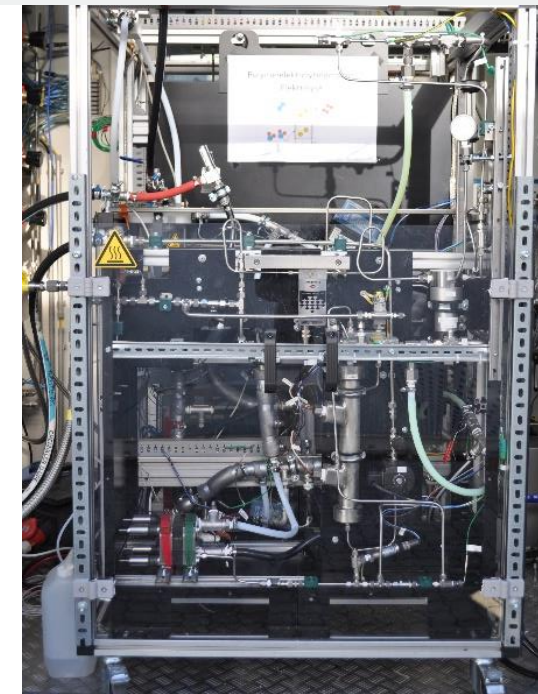
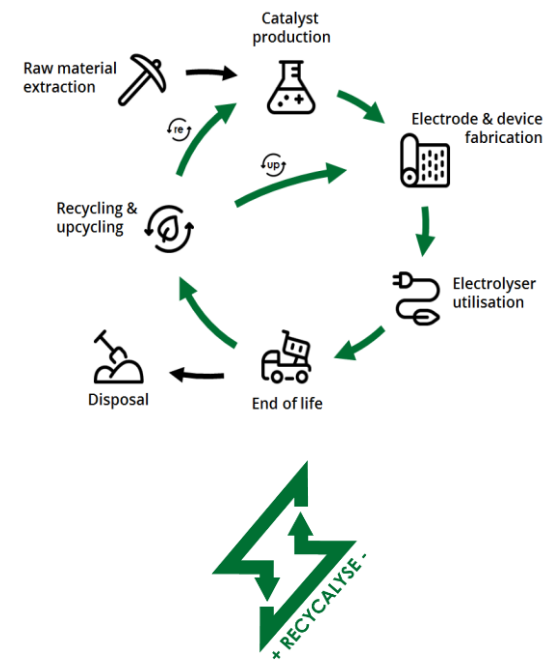
700 - 1000 °C

PEMEC: Iridium - Project Recycalyse

RECYCALYSE aims to develop new **electrocatalysts** for PEM electrolyser systems with increased **performance**, reduced **critical raw material** usage, red. environmental **footprint** and red. total **costs**.

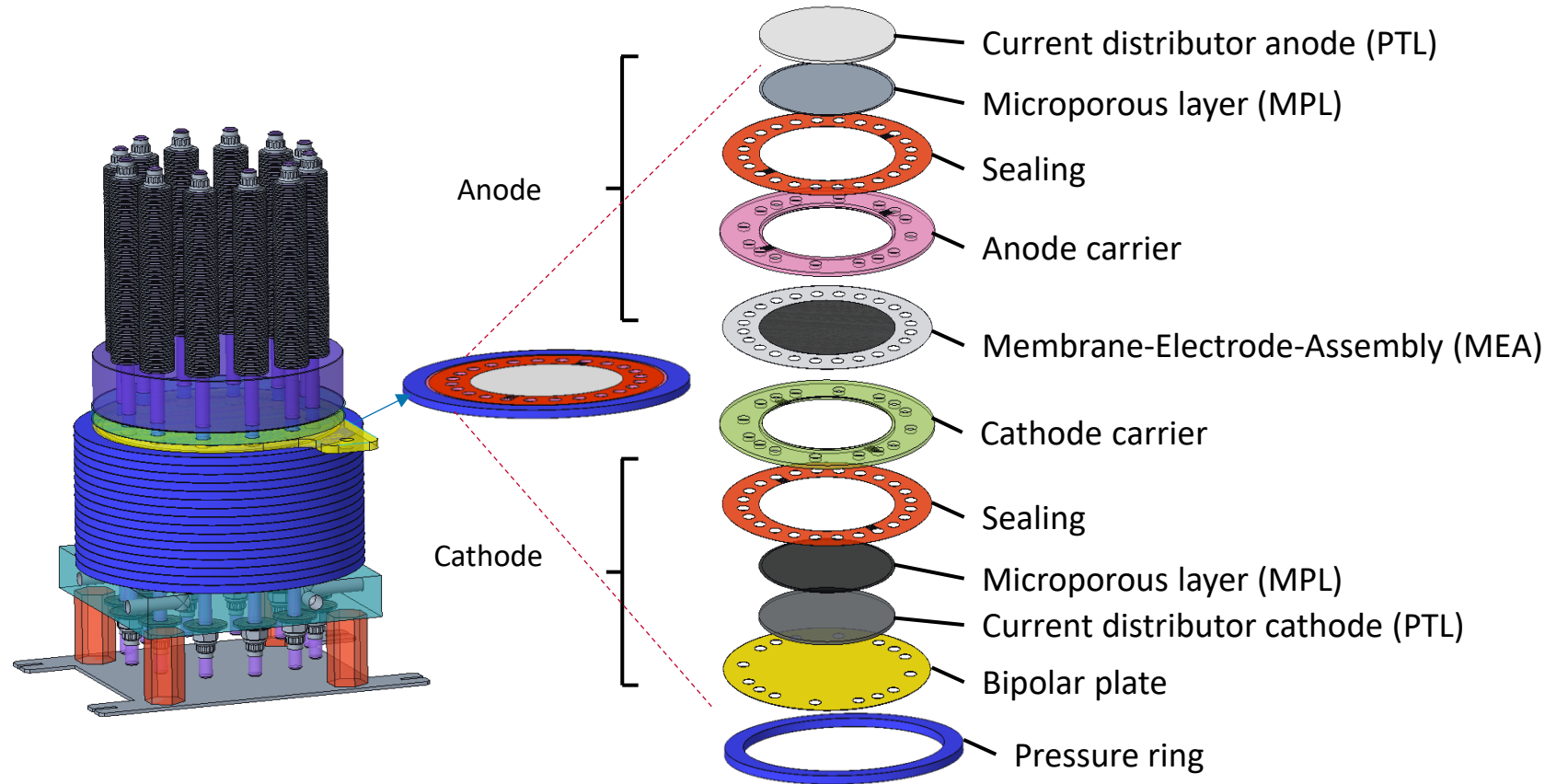
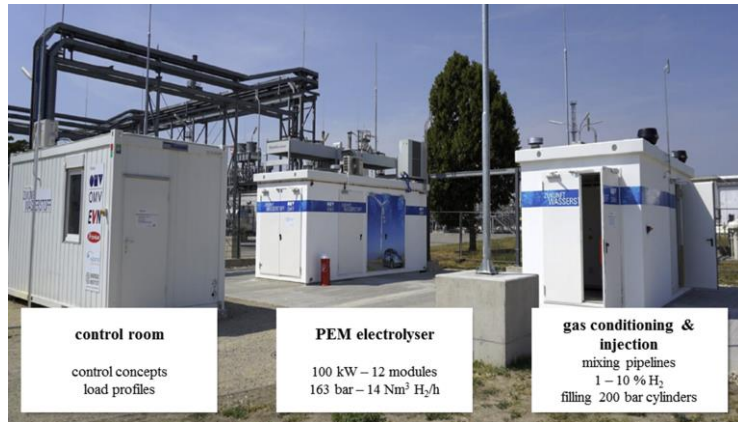
See more: Dr. Julia Melke et al., Recycalyse – New Sustainable and Recyclable Catalytic Materials for Proton Exchange Membrane Electrolysers, <https://doi.org/10.1002/cite.202300143>

- Process development for large scale recycling of the critical raw materials
- Application of sustainable materials derived from earth abundant elements
- Implementation of a circular economy in which the CRM will be recovered and regenerated
- Analysis of the entire value chain from catalyst manufacturing to system integration and demonstration, end-of-life recycling and supply of raw materials for the catalyst manufacturing



PEMEC: High Pressure

Wind2Hydrogen 163 bar HP-PEMEC



New Stack Development in COMET Programm

See more: Sartory, M., Wallnöfer-Ogris, E., Salman, P., Fellingner, T., Justl, M., Trattner, A., Klell, M.: "Theoretical and Experimental Analysis of an Asymmetric High Pressure PEM Water Electrolyser up to 155 bar", International Journal of Hydrogen Energy, 2017. <https://doi.org/10.1016/j.ijhydene.2017.10.112>

1. Kosteneffizienz & Materialien

- Einsatz günstigerer Katalysatoren möglich (z. B. Nickel statt Platin)
 - Langlebige, stabile Membranen ohne teure Rohstoffe noch in Entwicklung

2. Skalierbarkeit & Integration

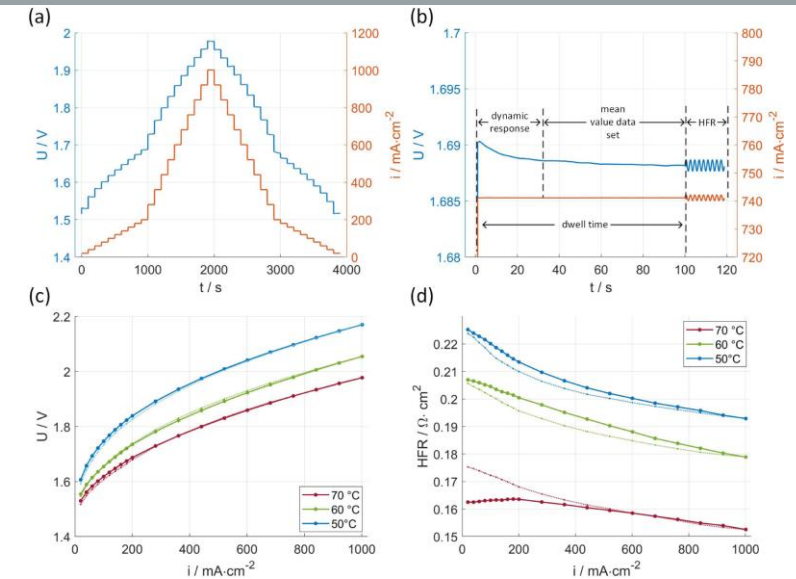
- Gute Eignung für dezentrale Anwendungen
 - Technische Standardisierung & Systemintegration noch ausbaufähig

3. Membranstabilität & Lebensdauer

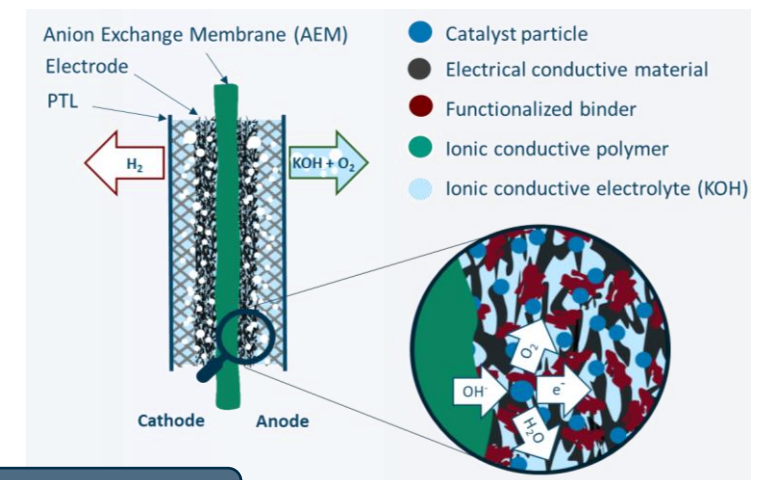
- Fortschritte bei chemisch beständigeren Polymeren
 - Aktuell geringere Lebensdauer im Vergleich zu etablierten Technologien

4. Wirkungsgrad & Betriebsführung

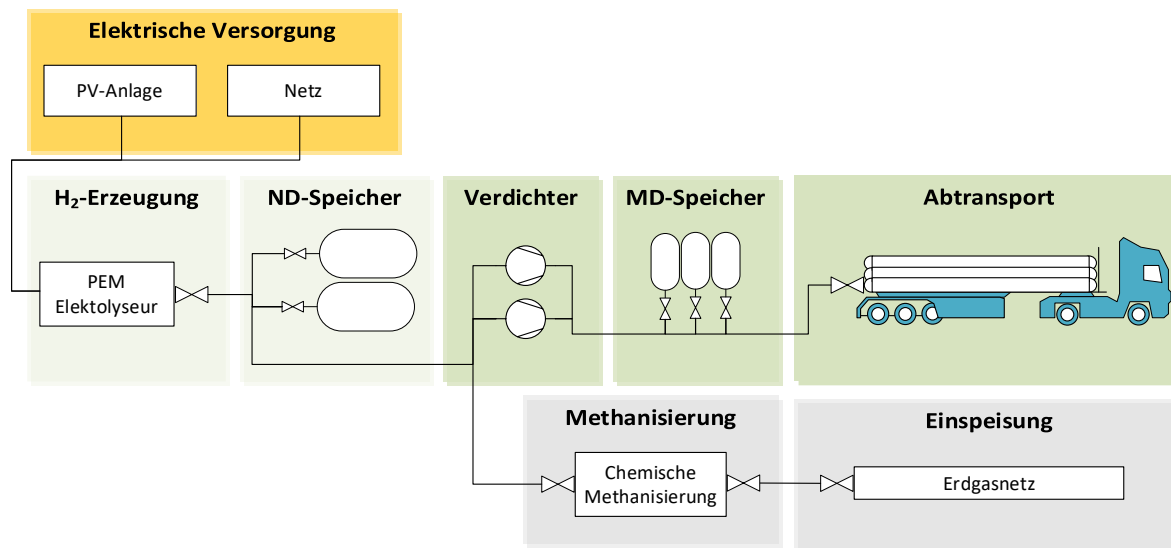
- Kombination der Vorteile von PEM und alkalischer Elektrolyse
 - Betriebsparameter noch nicht optimal für Langzeitanwendung



M. Ranz, B. Grabner, B. Schweighofer, H. Wegleiter, A. Trattner; Journal of Power Sources, 2024, <https://doi.org/10.1016/j.jpowsour.2024.234455>

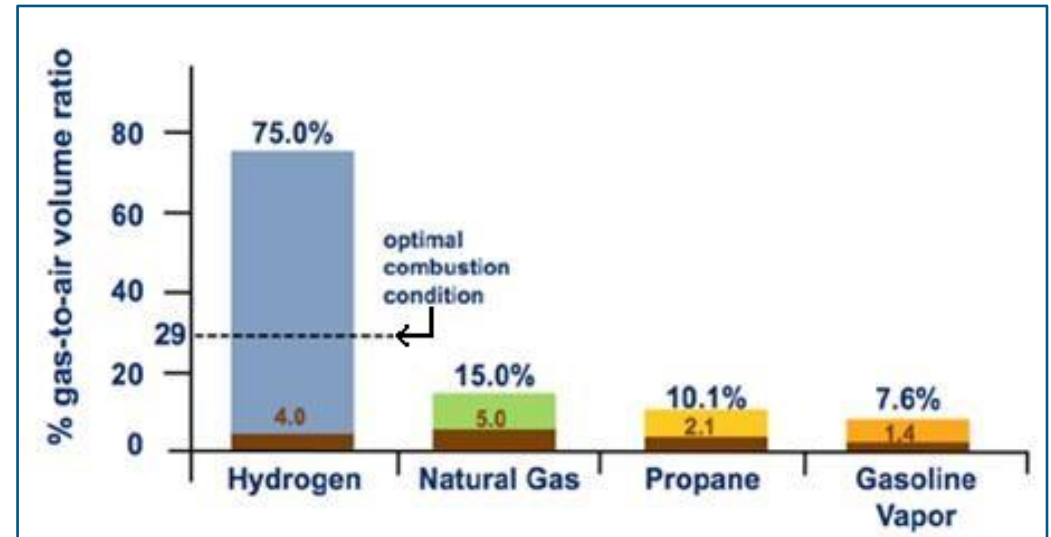


- Production of green hydrogen by using electricity from a local photovoltaic plant
- 1 MW Power-to-Gas Plant (PEM-Electrolysis)
- Filling the produced hydrogen into trailer (on site) and transferred to Wolfram Bergbau und Hütten AG
- Demonstration of the load-flexible methanation process
- Production of green biomethane and storage in the existing natural gas grid

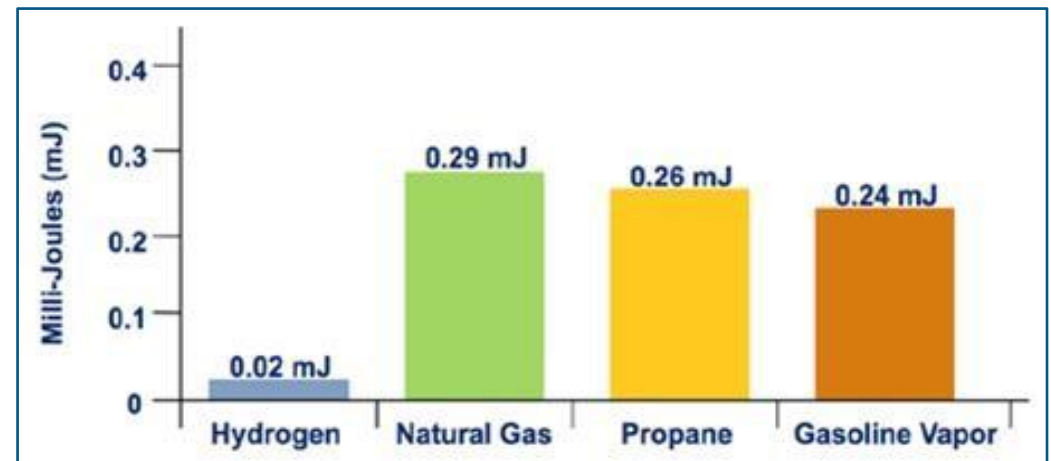


Sicherheitsrelevante Eigenschaften Wasserstoff:

- Farbloses, geruchloses Gas
- Keine toxischen Effekte
- Geringste Dichte aller Gase (14-mal leichter als Luft)
- Hohe Diffusionsneigung
- Niedrige Schmelz- und Siedetemperatur
- Leicht entzündbar
- Bildet mit Luft in einem weitem Mischungsbereich zündfähige Gemische mit hoher Flammgeschwindigkeit und hoher Verbrennungstemperatur
- Kennzeichnung/Gefahrenhinweise Wasserstoff, verdichtet
 - H220: Extrem entzündbares Gas
 - H280: Enthält Gas unter Druck; kann bei Erwärmung explodieren



Explosionsfähige Atmosphären im Vergleich [1]



Zündenergien [2]

Quelle: [1] <https://h2tools.org/hydrogen-compared-other-fuels>

[2] Klell, Eichlseder, Trattner – Hydrogen in Automotive Engineering, 2023

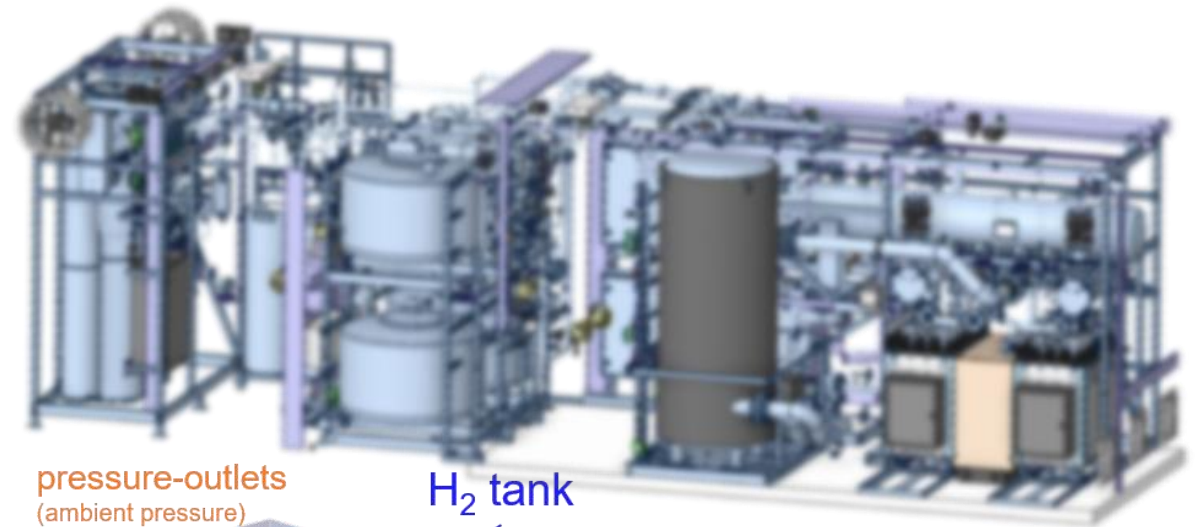
H₂-Austritt im Störfall

Evaluierung des Gefahrenpotentials eines 2,5MW Elektrolyseurs im Regelbetrieb

- Bewertung des Gefährdungspotentials aufgrund plötzlicher Wasserstoffleckage im Elektrolysebehälter
- Verteilung von Wasserstoff im Falle eines Rohrbruchs
- Wirksamkeit der Sicherheitsausrüstung und Überwachung
- Auswirkungen der Druckwelle
- Abschätzung der Explosionsintensität (TNT-Äquivalent)

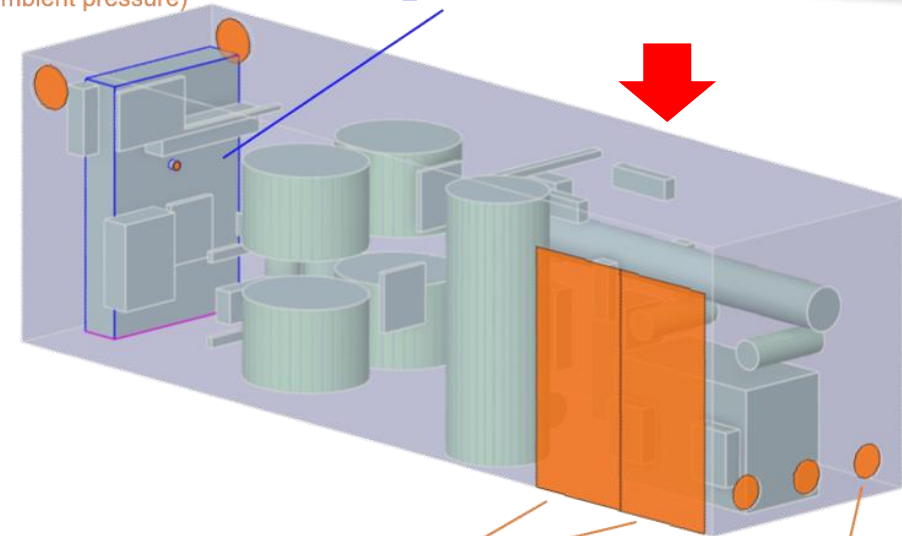
Worst-Case-Scenario:

- System im Regelbetrieb, H₂- Sensorik und Belüftung aktiv
- 35 bar aufgrund eines plötzlichen Wasserstofflecks (Rohrbruch ohne Rissbildung) mit 80 mm kreisrunder Bruchfläche
- 3kg H₂ max. Freisetzungspotential treten aus



pressure-outlets
(ambient pressure)

H₂ tank



wall / pressure-outlet
(switches when opening force is applied,
has ambient pressure when door is opened)

pressure-inlets
(Pressure-profile according
to stationary simulation)

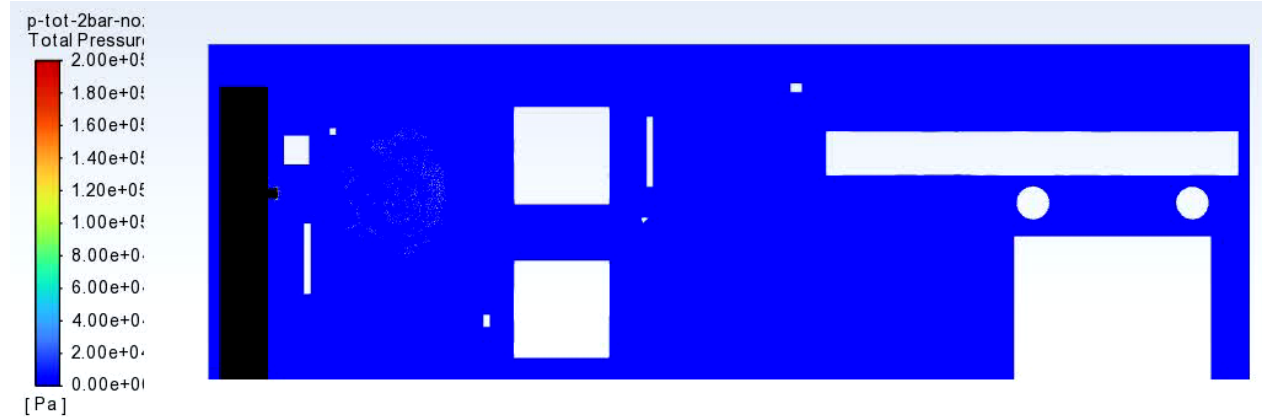
H₂-Austritt im Störfall

- Stationäre Belüftung kann die Bildung eines Ex-Gemischs nicht verhindern
- 65% des gesamten H₂-Freisetzungspotentials als Ex-Gemisch im Raum
- Explosion höchst wahrscheinlich
- Dabei wird Container deformiert aber nicht zerstört
- Druckwelle öffnet Explosionsklappen am Elektrolyse-Container und tritt durch diesen hindurch
- Druckwelle schleudert die Türen auf und kann Personen verletzen

Kernergebnisse:

- Mindestabstände zwischen Elektrolyseuren angepasst
- Zugangsbeschränkung bei Produktion

Druckwelle

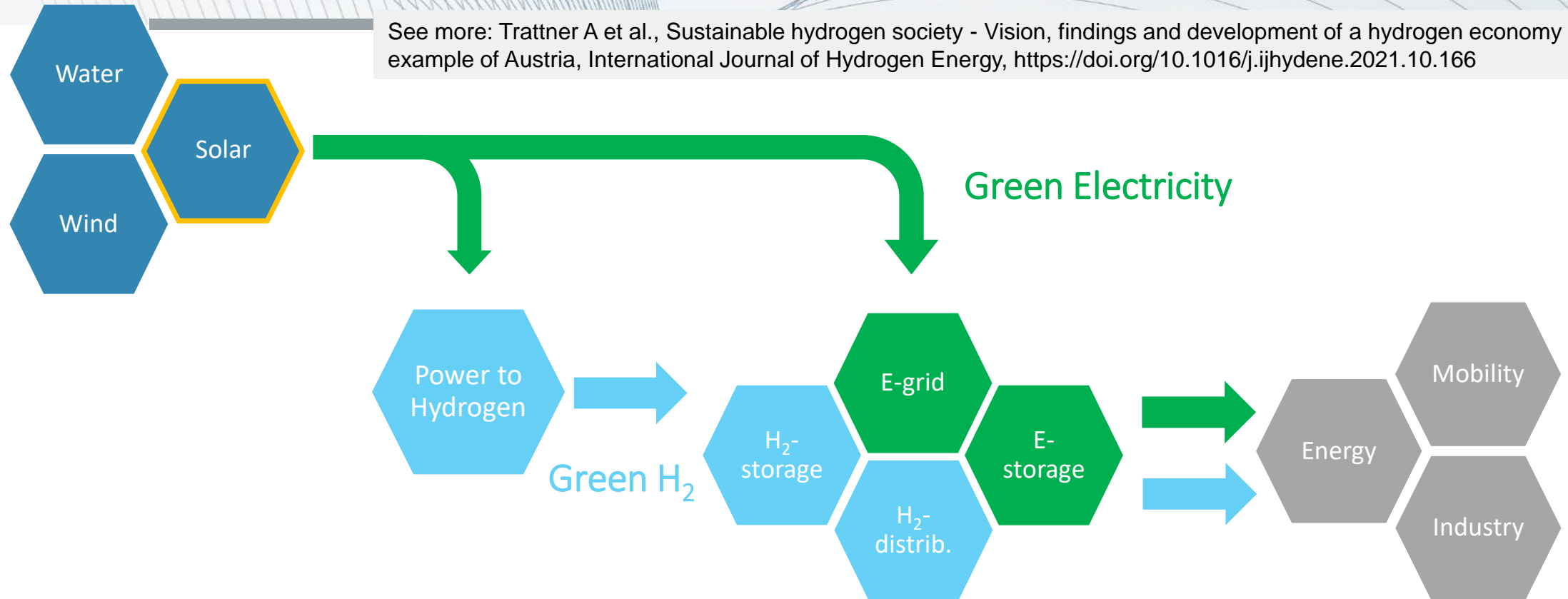


H₂ Ausbreitung im Raum



Hydrogen – a Key to the Energy Transition

See more: Trattner A et al., Sustainable hydrogen society - Vision, findings and development of a hydrogen economy using the example of Austria, International Journal of Hydrogen Energy, <https://doi.org/10.1016/j.ijhydene.2021.10.166>



Integration of renewables

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Energy conversion

- Electrolysis - compensate temporal volatility
- H₂ as secondary energy carrier – energy storage

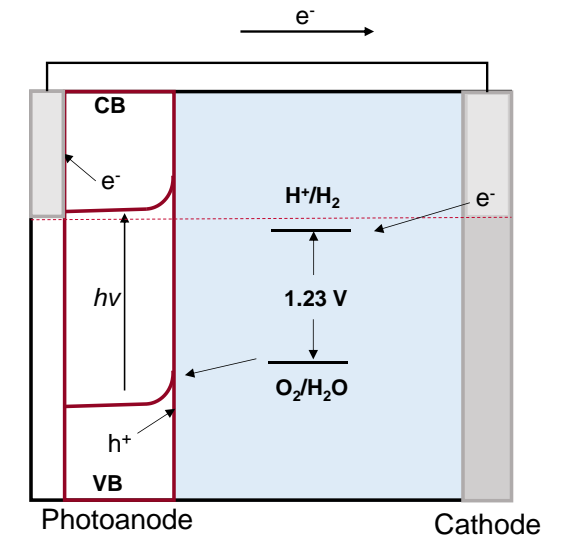
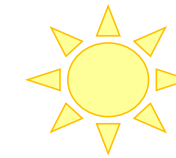
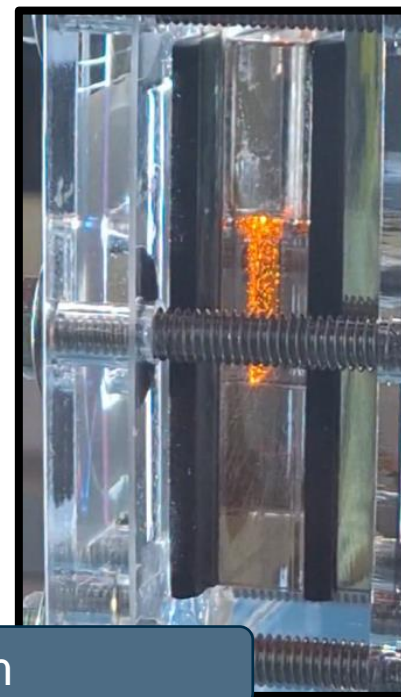
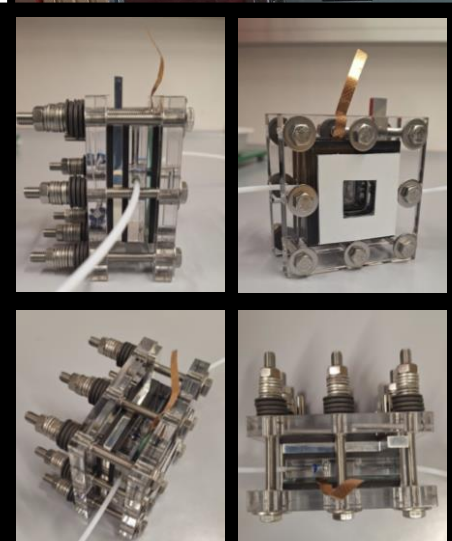
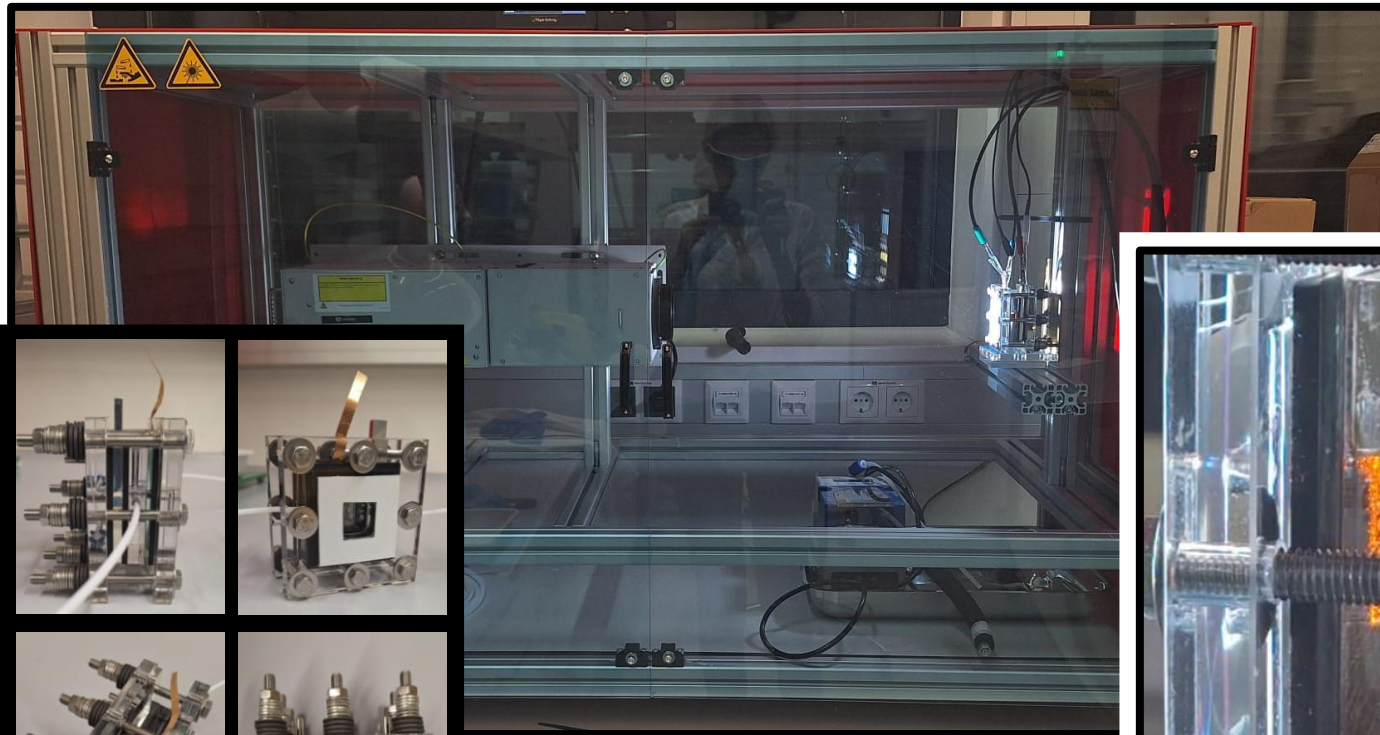
Storage and distribution

- Centralized and decentralized storage
- Long-term storage
- Efficient transport over long distances

Zero Emission Usage

- Energy Services – CHP
- Mobility with Fuel cells
- Industry and high-temperature processes

PEC prototype and test rig



Hisatomi, T.; Kubota, J.; Kazunari Domen, K.; Recent advances in semiconductors for photocatalytic and photoelectrochemical water splitting. *Chem. Soc. Rev.*, **2014**, *43*, 7520-7535; DOI: 10.1039/c3cs60378d

Lin, Y.-C.; Wyźga, P.; Macyk, J.; Macyk, W.; Guzik, M. N. Solar-driven (photo)electrochemical devices for green hydrogen production and storage: Working principles and design. *Journal of Energy Storage* **2024**, *82*, 110484. DOI: 10.1016/j.est.2024.110484.

Jiang, C.; Moniz, S. J. A.; Wang, A.; Zhang, T.; Tang, J. Photoelectrochemical devices for solar water splitting - materials and challenges. *Chemical Society reviews* **2017**, *46* (15), 4645–4660. DOI: 10.1039/c6cs00306k.

New Development in COMET Programm

Jonas Wolfsberger, Sarah Holler, Sarah Kurakin, Illena Grimmer and Marie Macherhammer

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